



In-situ stabilization of potentially toxic elements in two industrial polluted soils ameliorated with rock phosphate-modified biochars[☆]

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ABSTRACT

The present study was aimed at determining the efficacy of rock phosphate (RP) 3% loaded in a green coconut shell, chicken manure, and vegetable waste to make green coconut-modified biochar (GMB), chicken manure modified-biochar (CMB), and vegetable waste-modified biochar (VMB) in the fixation of Cr, Pb, Cu, Zn, Ni, and Cd in Sharafi goth and Malir polluted soils. The impact of RP impregnated with organic waste material to produce modified biochars (MBs) on stabilizing PTEs from polluted soils and reducing their uptake by mustard plant has not yet been thoroughly investigated. All modified BCs in 0.5, 1, and 2% doses were used to stabilize Cr, Pb, Cu, Zn, Ni, and Cd in two polluted soils and to reduce their uptake by the mustard plant. The obtained results revealed that the maximum mustard fresh biomass was 17.8% higher with GMB 1% in Sharafi goth polluted soil and 25% higher with VMB 0.5% in Malir polluted soil than in the control treatment. After applying modified BCs, immobilization of Cr, Pb, Cu, Ni, and Cd was observed in both soils and it reduced the uptake of these elements by mustard plants. On the other hand, although Zn mobilization increased by 0.38% for CMB 0.5% and by 5.9% for VMB 0.5% in Sharafi goth polluted soil, as well as by 3.15% for GMB 1%, 6.34% for GMB 2%, and 4.78% for VMB 0.5% in Malir polluted soil, this was due to changes in soil pH and OM. It was found that GMB 1%, CMB 0.5%, and VMB 0.5% have the potential to increase Zn uptake by mustard, while VMB 2% can reduce the element uptake by the plant. Redundancy analysis showed that soil chemical parameters were negatively correlated with PTEs in both soils and reduced their uptake by mustard. The present study revealed that MBs can stabilize PTEs in industrial and wastewater soils polluted with multiple metals and reduce their uptake by plants.

1. Introduction

Soil pollution with potentially toxic elements (PTEs) is considered one of the most severe ecological threats, while among reclamation approaches, investigation of soil additives has received significant attention (Wang et al., 2018; Rajendran et al., 2022). Excessive accumulation of PTEs such as Cr, Pb, Cu, Zn, Ni, and Cd in the soil is a global challenge due to geogenic and anthropogenic processes (Rinklebe and Shaheen, 2015; Palansooriya et al., 2020). Remediation of polluted soils

is imperative to ensure safer crop production (A. H. Lahori et al., 2020). Soil PTEs are considered the most polluting elements in soil, and their basic characteristics are non-degradability, perseverance, and bio-accumulation in the food web (Gholami et al., 2020). Nearly 12.6 million people have died worldwide in recent years from more than 100 diseases caused by unhealthy ecosystems, such as polluted soils (WHO, 2016). Soil pollution with PTEs leads to land degradation or loss of some soil functions on a global scale (Puga et al., 2015).

PTEs are released into the environment through atmospheric deposition from industrial effluent, dumping, mining, smelting, pesticides,

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List of abbreviations

PTEs	Potentially toxic elements
RP	Rock phosphate
GMB	Green coconut-modified biochar
CMB	Chicken manure-modified biochar
VMB	Vegetable waste-modified biochar
BCs	Biochars
MBs	Modified biochars
MLA	Malir industrial area
SGK	Sharafi goth korangi

sewage sludge, electric waste, commercial fertilizers, etc. (Qiao et al., 2020). There are more than 6000 different industrial plants in Karachi including refineries, chemical plants, petrochemical plants, tanneries, textile plants, pharmaceutical plants, solid waste and sewage sludge treatment plants, etc. All of them produce large amounts of effluent that is highly contaminated by organic and inorganic pollutants, which are also responsible for a significant degree of soil contamination (Afzal et al., 2018). Excessive accumulation of PTEs in soil leads to land degradation and retards crop production and requires remediation using environmentally friendly, feasible, and low-cost techniques (Derakhshan-Nejad and Jung, 2017). At the same time, PTE pollution in the food chain is a serious problem that threatens living organisms (Antoniadis et al., 2017). The mobility and adsorption of PTEs in soils are strongly influenced by the chemical fractionation of elements in the polluted soil (Chao et al., 2018). Numerous in-situ and ex-situ soil reclamation/restoration technologies have been used to reduce the risk of pollution with PTEs and exploit arable land for agricultural purposes to ensure food security (Wang et al., 2015; Beiyuan et al., 2017a,b). In-situ technologies viz., solidification/stabilization (Li et al., 2018), soil washing/flushing (Beiyuan et al., 2018), bioremediation (Park et al., 2011), and ex-situ techniques such as soil washing (Beiyuan et al., 2017a,b), vitrification landfilling, and bioreactors are often used to diminish bioavailability of PTE pollutants (Kuppusamy et al., 2016). With the growing threat to human health posed by polluted soils, the identification and optimization of soil restoration technologies are becoming increasingly important. All these soil restoration technologies have dissimilar overall working values and specific merits and demerits. In addition, technology potential and cost-effectiveness can vary drastically between in-situ and ex-situ trials (Khalid et al., 2017). Among these techniques, the application of soil additives to stabilize PTEs has been extensively employed to restore PTE-polluted soils; this results from its rapid and easy incorporation as well as commercial feasibility (Beiyuan et al., 2016). Choosing the appropriate stabilizers can deliver cost-effective restoration methods and fulfill the “green and sustainable remediation” assumption due to their lower life cycle ecological footprint (Hou and Al-Tabbaa, 2014).

Certainly, rock phosphate (RP) is the primary source of P that can be used to manufacture single superphosphate (SSP) or triple superphosphate (TSP) fertilizer in Pakistan to improve soil health and crop production (Lahori et al., 2019). Involvement in the production of phosphate raw materials with a P_2O_5 content less than 24% is not economically feasible. The dominant phosphate reserves are located in Khyber Pakhtunkhwa and Hazara divisions in the northeastern part of the country; unfortunately, there are no consistent skills to make it suitable for local alkaline soils. The soils of Pakistan are alluvial, with $CaCO_3 > 3.0\%$ and no native P. Indeed, RP is a natural, inexpensive, and clean complex material, but, unfortunately, has a low P content and its solubility is too slow to meet the plant demand and stabilize pollutants. As many as 90% of soils in Pakistan are already deficient in P. Direct application of RP to acidic soil may solubilize the P content, whereas its solubility may be impaired in alkaline and/or calcareous soils due to

high pH and low organic matter content (Ishtiaq and Ahmad, 1987; Rahman et al., 2018). It can be assumed that mixing PR with organic waste material to produce modified BCs can be highly effective in stabilizing PTEs in multi-metal-polluted soil and reduce metal uptake by plant, as PR has a high potential to adsorb pollutants in the soil medium (Caravaca et al., 2004). Biochar (BC) as an amendment originated in Terra Preta Soils (Glaser et al., 2002). In the last few years, BC has been broadly studied and applied as an efficient soil development and remediation material (Downie et al., 2009; Ahmad et al., 2017). BC is a solid and C-rich product formed by heating feedstock under O_2 -limited conditions (Cao and Harris, 2010). Generally, BC is pyrolyzed at low and high temperatures from agricultural/forestry waste, fruit waste, algae, tire waste, dead animals, food waste, and livestock manure as a feedstock for PTE remediation in polluted soils (Novak et al., 2013; Gul et al., 2015). Several studies have shown that the BC can reduce the bioavailability of PTEs in soil by increasing soil pH, cation exchange capacity, carboxyl groups, hydroxide ion complexation, etc. (Lahori et al., 2017; Igalavithana et al., 2017). Some researchers have found that BC can increase nitrogen (N) content, water holding capacity, nutrient retention, microbial activity, soil fertility and productivity (Houben et al., 2013; Puga et al., 2015). Over the past few years, modified biochar (MBC) has been successfully used to stabilize PTEs in polluted soils. A study by Ren et al. (2020) assessed the potential of $Ca(H_2PO_4)_2$ -based pig manure biochar to bind Cd and Pb in smelter-polluted soils. Furthermore, Gholami et al. (2020) examined the efficacy of thiourea-modified BC in the stabilization and speciation of Cd and Pb in acidic polluted soil. Additionally, Irshad et al. (2020) investigated the potential of goethite-modified biochar to fix Cd and As in contaminated soil and promote the growth of rice. Zhang et al. (2020) applied phosphorus-modified biochars to stabilize Cu, Cd, and As in contaminated paddy soil. What is more, Wang et al. (2021) reported that Fe/Mn- and P-modified drinking water treatment residuals can decrease Cu and Pb concentrations in mining-polluted soil and reduce their uptake by ryegrass. From a novel point of view, limited information is available on the activation of 3% rock phosphate impregnated with green coconut shell, chicken manure, and vegetable waste biomass to make modified biochars (MBs) pyrolyzed at $<500\text{ }^\circ\text{C}$ to stabilize Cr, Pb, Cu, Zn, Ni, and Cd in industrial soils polluted with multiple metals and reduce their absorption by mustard plants. Therefore, the main objectives of the present study were: 1) to explore the impact of rock phosphate loaded 3% in green coconut shell, chicken manure, and vegetable waste to make green shell-modified biochar (GMB), chicken manure-modified biochar (CMB), and vegetable waste-modified biochar (VMB) on the fixation of Cr, Pb, Cu, Zn, Ni, and Cd in Sharafi goth and Malir polluted soils; 2) to investigate the effect of different concentrations of modified biochars on the phytoavailability of Cr, Pb, Cu, Zn, Ni, and Cd to the mustard plant; and 3) to assess the impact of different rock phosphate-modified biochars on soil chemistry. It was hypothesized that different rock phosphate-modified biochars can immobilize multi-toxic elements in industrial and wastewater polluted soils and reduce their uptake by plants.

2. Material and methods

2.1. Soil collection and characterization

In the present study, two different soils samples were collected from Malir Industrial Area (MLA) ($24^\circ 51'38''\text{N}$; $67^\circ 13'30''\text{E}$), Sharafi goth Korangi (SGK) ($24^\circ 85'76''\text{N}$; $67^\circ 16'76''\text{E}$) Sindh Province, Pakistan (Fig. S1). Surface soil samples were collected from 0 to 15 cm layer from Sharafi goth and Malir areas. Composite soil samples were taken from different points. The soil samples were air-dried for 1 week. Debris and non-soil material were separated manually, and soil samples were $<2\text{ mm}$ ground. Selected basic physicochemical parameters of both soils are given in (Table 1).

Table 1
Physico-chemical characteristics of the studied soil and its amendments.

Parameters	SGPS	MPS	RP	GMB	CMB	VMB
Soil texture	Sandy loam	Sandy clay loam	-	-	-	-
pH	6.99 ± 0.7	6.92 ± 0.2	7.80 ± 0.8	7.95 ± 0.7	8.44 ± 0.4	10.6 ± 0.6
EC (mS/cm)	2.37 ± 1.2	2.05 ± 0.6	0.9 ± 0.2	27.6 ± 1.2	42.3 ± 0.5	66.6 ± 0.1
SOC (g/kg)	40.9 ± 0.3	40.1 ± 1.2	-	49.9 ± 1.8	55.3 ± 0.2	66.53 ± 0.2
Organic matter (g/kg)	70.6 ± 1.2	69.3 ± 0.7	-	86.1 ± 0.9	96.08 ± 0.8	100.2 ± 0.6
Pb (mg/kg)	70.7 ± 1.6	143.6 ± 0.9	-	-	-	-
Cd (mg/kg)	8.3 ± 0.6	11.4 ± 1.5	-	-	-	-
Zn (mg/kg)	470.1 ± 0.5	483.8 ± 1.1	-	-	-	-
Cu (mg/kg)	38.9 ± 1.2	46.6 ± 0.7	-	-	-	-
Cr (mg/kg)	239.8 ± 1.5	246.9 ± 1.4	-	-	-	-
Ni (mg/kg)	9.9 ± 0.5	15.3 ± 0.7	-	-	-	-
Biochar yield (g/100 g dry matter)	-	-	-	37.8 ± 0.6	51.02 ± 0.8	33.4 ± 0.3

SGPS: Sharafi Goth polluted soil.
MPS: Malir polluted soil.
RP: Rock phosphate
GMB: Green coconut shell-modified biochar
CMB: Chicken manure-modified biochar
VMB: Vegetable waste-modified biochar

2.2. Material collection and modified biochar preparation

Rock phosphate (RP) was obtained from Kakul Mine Haripur Hazara areas of Khyber Pakhtunkhwa (34° 10' 05" N; 73° 13' 28" E). Green coconut shell was collected at the local market, chicken manure came from chicken farms, and vegetable waste was collected at the vegetable market in Karachi, Pakistan. In the first step, to prepare modified BCs, dry feedstocks, i.e. green coconut shell, chicken manure, and vegetable waste, were ground and 2 mm sieved, and then <1 mm sieved dry rock phosphate (RP) was carefully mixed with individual feedstocks. In the second step, all the mixed material was placed in a muffle furnace and pyrolyzed at <500 °C for 3 h to produce green coconut-modified biochar (GMB), chicken manure-modified biochar (CMB), and vegetable waste-modified biochar (VMB). Except for the control treatment, all three MBs were added to both soils in 0.5, 1, and 2% doses.

2.3. Experimental set-up

The efficacy of modified biochars in the phytoextraction of Ni, Cr, Pb, Cu, Zn, and Cd by mustard (*Brassica napus*) was investigated on Sharafi goth and Malir polluted soils in a greenhouse pot experiment. Immobilizing agents such as green coconut-modified biochar (GMB), chicken manure-modified biochar (CMB), and vegetable waste-modified biochar (VMB) were applied in doses of 0.5, 1, and 2% (w/w) (Ali et al., 2019). In addition, all modified biochars were ground, <2 mm sieved, and carefully mixed with polluted soils, except for the control soil samples. Approximately 1.0 kg of each air-dried sample together with 0.5, 1, and 2% doses (dry matter/weight) of each modified BC were placed in 38 cm diameter pots to cover each pot to a height of 12 cm. All treatments were applied to polluted soils in three replicates – randomized complete block design (RCBD). This design included 60 pots (2 soils × 10 treatments × 3 replicates and 1 crop). Soils and each modified BC were thoroughly mixed and placed in pots for investigation. All pots were irrigated with almost 300 ml of tap water to achieve a moisture

content of approximately ~65% of the soil moisture content. All pots were incubated for 15 days for chemical reaction. Furthermore, mustard seeds were treated with a suspension of H₂O₂ (3%, v/v) for 15 min and then washed with pure water about 3 times. About 5 to 7 seeds were sown in each pot and after 10–15 days, reduced to 3 plants per pot. At the time of mustard sowing, the field capacity was adjusted to nearly 80%. Water losses in each pot were replenished daily. All mustard plants were harvested within 50 days.

2.4. Sample analysis

2.4.1. Soil and modified biochar analyzes

All chemical solutions were prepared using deionized water, and all chemicals used in our study were of pre-analytical grade. The EC and pH were measured using deionized water (1:5 H₂O ratio) in both soil samples, rock phosphate, and modified biochars, respectively. The SOC content in soils and modified BCs was determined using elemental analysis (Mebius, 1960). The total lead, cadmium, copper, chromium, zinc, and nickel contents in modified BCs were tested by the United States Environmental Protection Agency method 3050 B (The United States Environmental Protection Agency, 1996). The surface functional groups of all three modified BCs were characterized by Fourier transform infrared (FTIR) spectroscopy with the use of a Bruker Vector 22 FTIR spectrophotometer in the range of 4000–400 cm⁻¹. The samples were mixed with KBr powder (1%, w/w) until homogenized and pressed by applying mechanical pressure to form a KBr disk (Fig S2).

2.4.2. Mustard plant analysis

Mustard plants were harvested about fifty days after planting. All plants were carefully uprooted, washed, and then placed in paper bags to determine total biomass (shoots and roots) and for further chemical analyzes. The dry matter of plant biomass (shoots and roots) was recorded after drying at 65 °C for 3–4 days. The dried plant biomass was then crushed in a small electric machine to obtain powder, after which the crushed dry biomasses of roots and shoots were stored in paper bags for testing. Almost 0.5 g of HNO₃ – HClO₄ mixture in a 4:1 ratio was used according to Lindsay (1978) to determine the contents of nickel, copper, lead, chromium, cadmium, and zinc in plant roots/shoots. The total contents of these elements in the treated samples were detected by digesting 0.2 g of soil in a mixed solution of HF–HClO₄– HNO₃– Zhu et al. (2018). The proportions of PTEs in the dissolved samples were detected by AAS instrument using the United States Environmental Protection Agency method 3050 B (The United States Environmental Protection Agency, 1996).

2.5. Statistical analysis and quality control

All trial data were statistically analyzed using Statistix 8.1. The mean of 3 replicates was subjected to the HSD test at p < 0.05. OriginPro 16 software was used to prepare the graphs. RDA analysis was performed in CANOCO 5 to determine the correlation between soil toxic metals, plant growth, toxic metals in plant roots and shoots, and soil chemical properties. Certified reference materials for soil GBW07457 (GSS-28) and plant GBW07603 (GSV-2 mustard) according to the Chinese Academy of Geological Sciences were used for quality control. The recovery rates of Pb, Cu, Zn, Ni, and Cd in soil were 93–101, 94–103, 95–102, 91–101, and 94–103%, respectively, and in plant 91–104, 93–102, 92–104, 90–101, and 93–105%, respectively.

3. Results and discussion

3.1. Properties of studied soils and additives

The properties of Sharafi goth and Malir soils are shown in (Table 1). Soil pH was nearly neutral (6.99 and 6.92) and the organic matter content was higher (70.12 g/kg and 69 g/kg), with moderate EC (2.37

and 2.05 mS/cm, respectively). The total concentration of PTEs was compared with standards (Denneman and Robberse, 1990; WHO, 1996). All the PTEs tested exceeded the limits of the WHO standard, with the exception of Ni, and were therefore compared with the standard value of 35 mg/kg in the studied experiment. The physicochemical results revealed that both soils were highly polluted with PTEs. In addition, the texture of Malir polluted soil was sandy clay loam, while in the case of Sharafi goth it was sandy loam (Table 1). The GMB and CMB additives had slightly neutral pH, while VMB was alkaline. The pH values of the additives were: 7.95 for GMB, 8.44 for CMB, and 10.6 for VMB. The electrical conductivity values were 27.6 mS/cm (GMB), 42.3 mS/cm (CMB), and 66.6 mS/cm (VMB). The soil organic carbon content was the highest in VMB (66.6 g/kg), followed by VMB (55.34 g/kg) and GMB (49.9 g/kg). The highest organic matter content was found in VMB (100.2 g/kg), followed by CMB (96.08 g/kg) and then GMB (49.9 g/kg). In terms of biochar yield, the most biochar was produced from CMB

(51.02%), followed by GMB (37.8%), and the least from VMB (33.40%).

3.2. FTIR results of modified biochars

The Fourier transform infrared (FT-IR) spectrum of biochar pyrolyzed from GMB, CMB, and VMB is illustrated in (Fig. S2). FTIR analysis was performed to evaluate the chemical components or functional groups present in the biochar, which are responsible for the stabilization of PTEs in contaminated soil. The red trace FTIR spectra revealed the presence of CMB, which showed a peak at 3450 cm^{-1} and 2922 cm^{-1} , corresponding to the O-H and C-H stretching frequencies, respectively. The peak that appeared at 1641 cm^{-1} indicated the carbonyl (C=O) of amide, while the peak at 1424 cm^{-1} revealed the bending frequency of N-H of the secondary amide. The medium intense peak at 1093 cm^{-1} indicated the presence of a C-O bond in the system. The black trace FTIR spectra showed the major peaks in the GMB biochar. The broad peak at

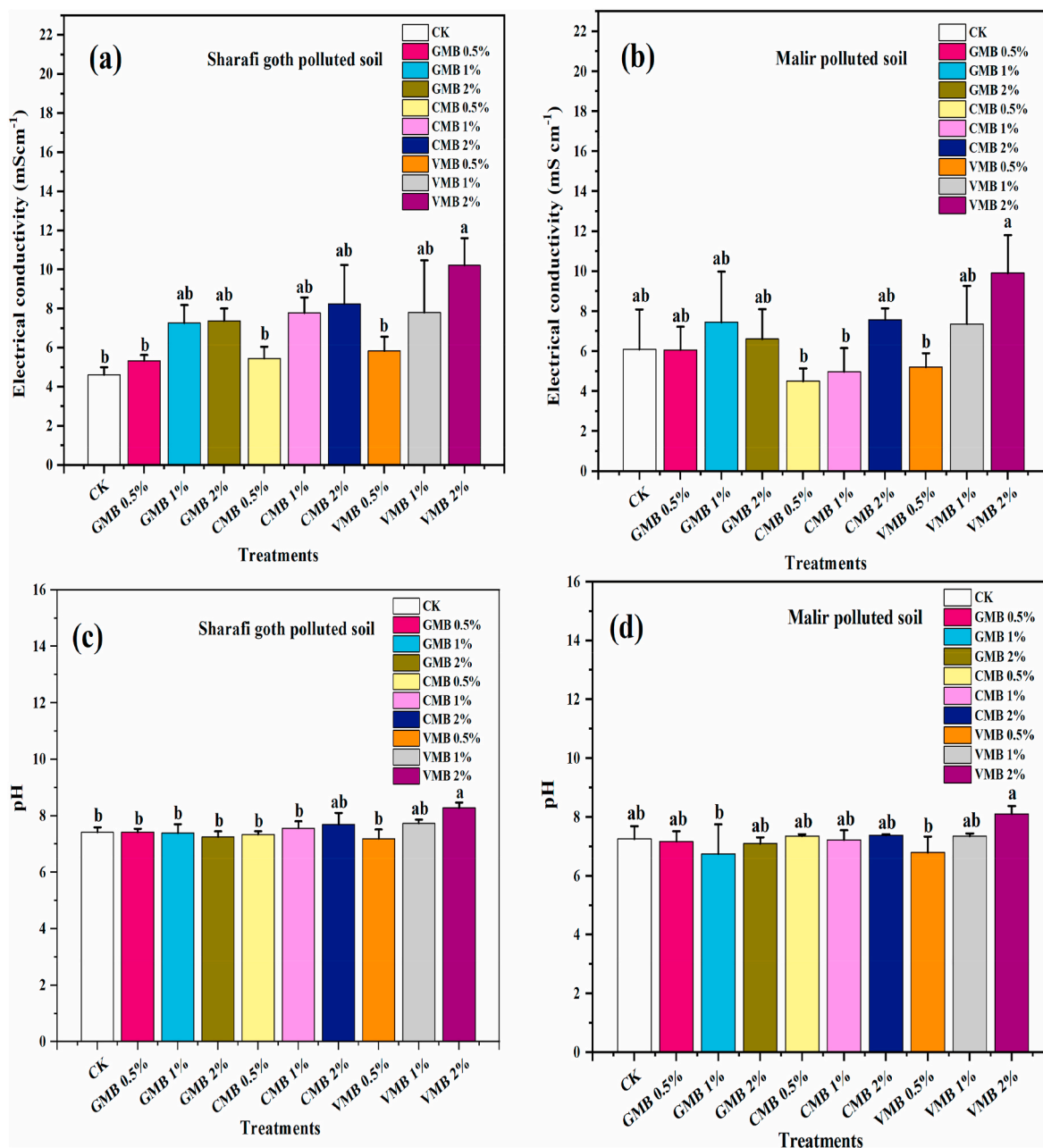


Fig. 1. Impact of modified biochars on: electrical conductivity in Sharafi goth polluted soil (a) and Malir polluted soil (b); pH in Sharafi Goth polluted soil (c) and Malir polluted soil (d).

3438 cm^{-1} of high intensity corresponded to the presence of O–H stretching vibrations with high hydrogen bonds and the N–H group. On the other hand, the absorption peak at 2923 cm^{-1} indicated the presence of asymmetric C–H stretching frequency. The peak at 1593 cm^{-1} was attributed to the vibration (C = C) and the peak at 1384 cm^{-1} to bending vibrations of $-\text{CH}_3$ and the C–N group. The peak at 1259 cm^{-1} corresponded to C–O stretching vibrations; finally, the peak at 671 cm^{-1} was attributed to C–H stretching vibrations and/or halogen trace as previously indicated in the literature (Alshehri et al., 2014). The blue trace FTIR spectra revealed the major peaks in the VMB biochar. The broad peak at 3550 to 3377 cm^{-1} corresponded to the presence of O–H stretching vibrations with hydrogen bonds and the N–H group. On the other hand, the absorption peak at 2923 cm^{-1} indicated the presence of asymmetric C–H stretching frequency. The absorption peak that appeared at 1741 cm^{-1} was attributed to the vibration (C = O) and the peak at 1399 cm^{-1} to bending vibrations of $-\text{N}-\text{H}$ and the C–N group. The peak at 1040 cm^{-1} corresponded to C–O stretching vibrations, while the peak at 671 cm^{-1} was attributed to halogen and/or C–H stretching vibrations of the aliphatic alkane (Fig. S2).

3.3. Effect of amendments on soil EC and pH

After harvesting mustard (*Brassica juncea* L) from pots, the impact of modified BCs on soil EC and pH were investigated in Sharafi goth and Malir polluted soils. All modified BCs significantly increased EC in Sharafi goth polluted soil, while for Malir polluted soil more diversified parameter changes were observed. The maximum EC in Sharafi goth polluted soil ranged from 4.62 to 10.22 mS/cm with the application of VBC 2% (Fig. 1a). In Malir polluted soil, the highest EC was between 6.10 and 9.91 mS/cm with the addition of VMB 2%, while the lowest EC ranged from 6.10 to 4.50 mS/cm with the application of CMB 0.5% compared to the control (Fig. 1b). The addition of BC ash into the polluted soil increased the EC of soil (Kolton et al., 2011). Ali et al. (2019) reported that EC and pH in mining soil increased after applying wood BC to it. The addition of biochars significantly increased electrical conductivity in calcareous soil Karimi et al. (2020).

Changes in pH levels were observed in Sharafi goth and Malir polluted soils. Compared to the control treatment, the highest pH level in Sharafi goth polluted soil ranged from 7.41 to 8.28 with the

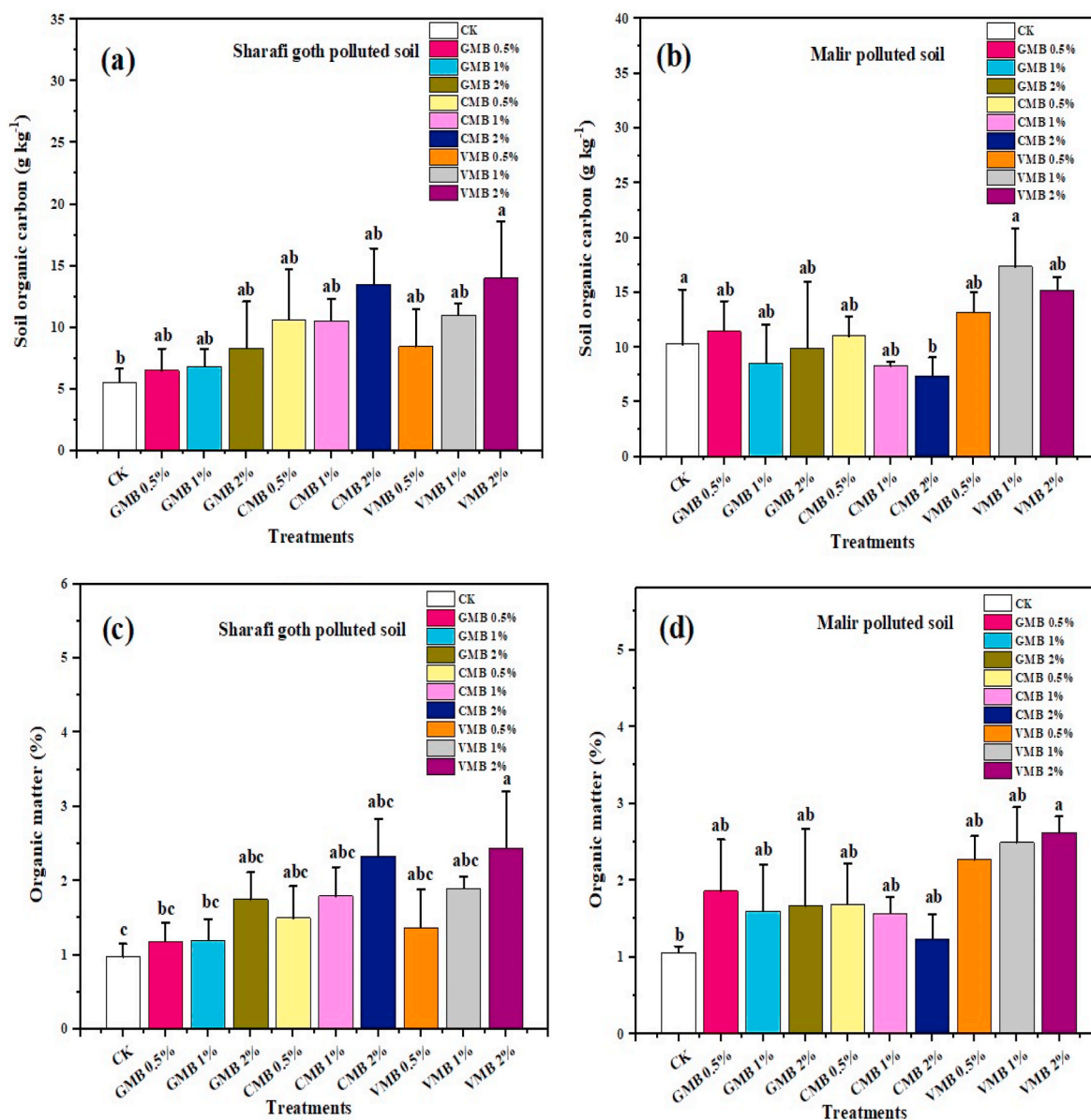


Fig. 2. Impact of modified biochars on: soil organic carbon in Sharafi Goth polluted soil (a) and Malir polluted soil (b); organic matter in Sharafi goth polluted soil (c) and Malir polluted soil (d).

application of VMB 2%, and the lowest pH level was between 7.41 and 7.19 following the addition of VMB 0.5% (Fig. 1c). The application of modified BCs caused changes in Malir polluted soil. The highest pH value significantly increased and ranged from 7.26 to 8.11, with the highest value reported after the application of VMB 2%. The lowest pH value was significantly reduced from 7.26 to 7.10 with the application of GMB 2% compared to the control soil (Fig. 1d). It should be noted that pH is the key factor through which biochars interact with pollutants (Ahmad et al., 2014). Higher metal removal efficiency with increasing BC application rate was attributed to an increase in pH (Xu et al., 2013; Komkiene and Baltreinaite, 2016). According to Berihun et al., 2017, soil pH can be improved by applying BC, by comparing it with the control samples of soil amendment. Zhang et al. (2019) stated that the application of BC and fertilizer can cause minor variation in soil pH. The application of BC in soil has the potential to increase soil pH, which can be considered one of the key factors for stabilizing PTEs in polluted soil (Taskin et al., 2019).

3.4. Effect of amendments on soil SOC and OM

All modified biochars were able to significantly increase the soil organic carbon (SOC) in Sharafi goth polluted soil as compared with control. The highest SOC observed in Sharafi goth polluted soil was 60.50% after applying VMB 2% (Fig. 2a). Changes in SOC were observed in Malir polluted soil after harvesting mustard plants grown with modified biochars. The maximum concentration of SOC in Malir polluted soil reached 40.8% after applying VMB 1%, while CMB 2% reduced the SOC content to 28.08% as compared with control (Fig. 2b). All modified biochars were found to be highly effective in increasing soil organic matter (OM) in Sharafi goth and Malir polluted soils. The maximum OM content in Sharafi goth and Malir soils was 60.24% and 59.6%, respectively, with the application of VMB 2% as compared with control (Fig. 2c and d). Karimi et al. (2020) reported that the application of BCs enhanced SOC in alkaline calcareous soil.

3.5. Impact of soil amendments on fresh and dry biomass yield

The growth of mustard plant in Sharafi goth and Malir polluted soils

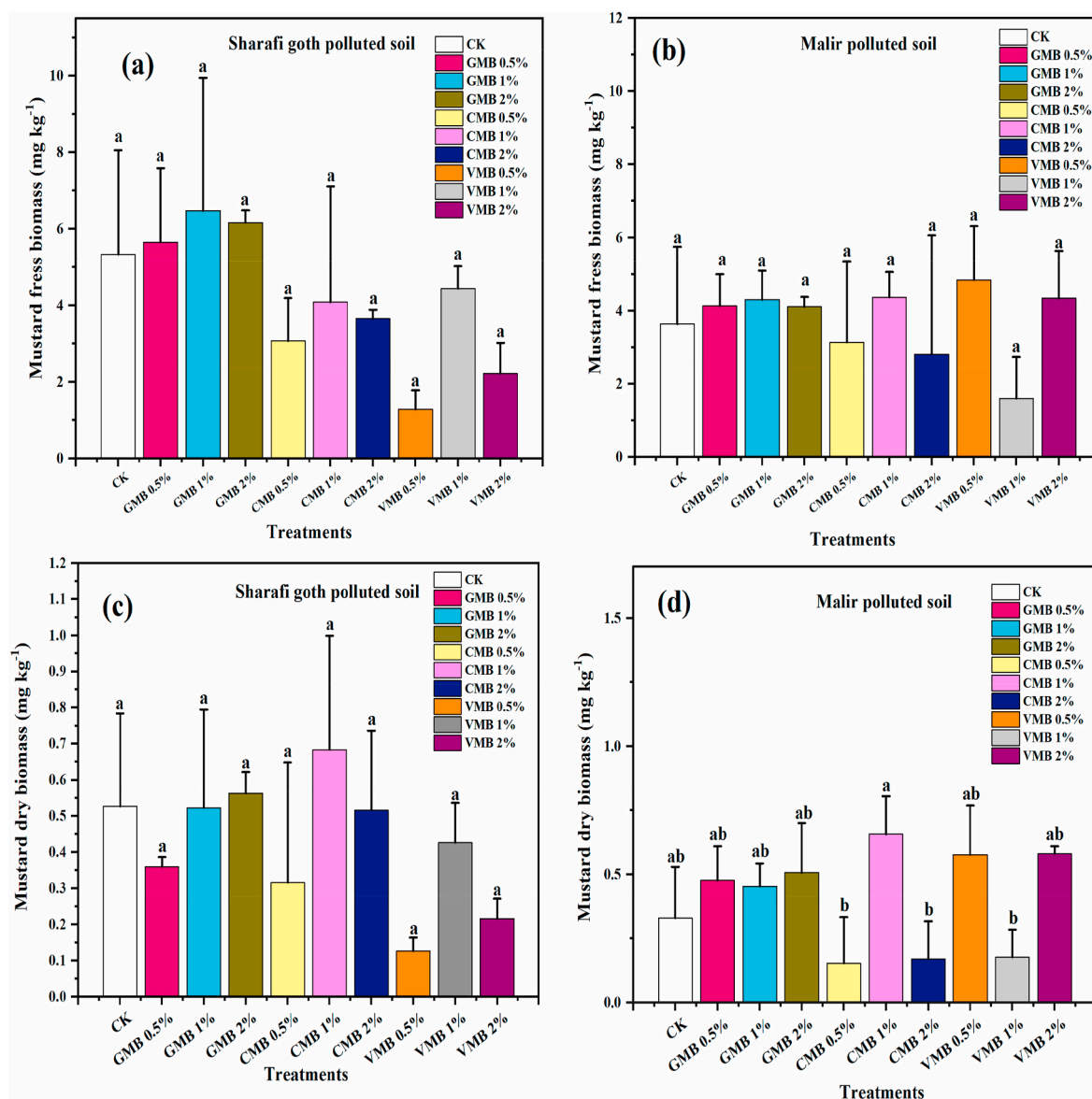


Fig. 3. Impact of modified biochars on: mustard fresh biomass in Sharafi goth polluted soil (a) and Malir polluted soil (b); mustard dry biomass in Sharafi goth polluted soil (c) and Malir polluted soil (d).

was higher after applying GMB in 0.5%, 1%, and 2% doses compared to other modified biochars. In addition, the application of GMB 1% in Sharafi goth polluted soil significantly increased the mustard fresh biomass, i.e. to 17.8%, whereas the lowest fresh biomass of mustard plant was noted by 75.98% with application of VMB 0.5% than control treatment (Fig. 3a). The greatest amount of fresh biomass of mustard plant was obtained with the application of VMB 0.5% and it accounted for 17.6% compared to the control. The application of VMB 1% to Malir polluted soil non-significantly reduced the fresh biomass of mustard plant to 17.6% as compared with control, followed by other modified

biochars (Fig. 3b). Mustard dry biomass in Sharafi goth polluted soil increased insignificantly by 22.02% after adding CMB 1%; however, VMB in a low dose evidently reduced mustard dry biomass to 75.4% compared to the control treatment (Fig. 3c). The maximum (50%) mustard dry biomass was observed after applying CMB 1%, while the minimum parameter value (54.7%) was recorded after introducing CMB 0.5% as compared to the treatment without any amendments (Fig. 3d). According to Khan et al. (2017), the application of vegetable waste- and chicken manure-modified biochars significantly increased the biomass yield of pak choi as compared to the control treatment. Ali et al. (2019)

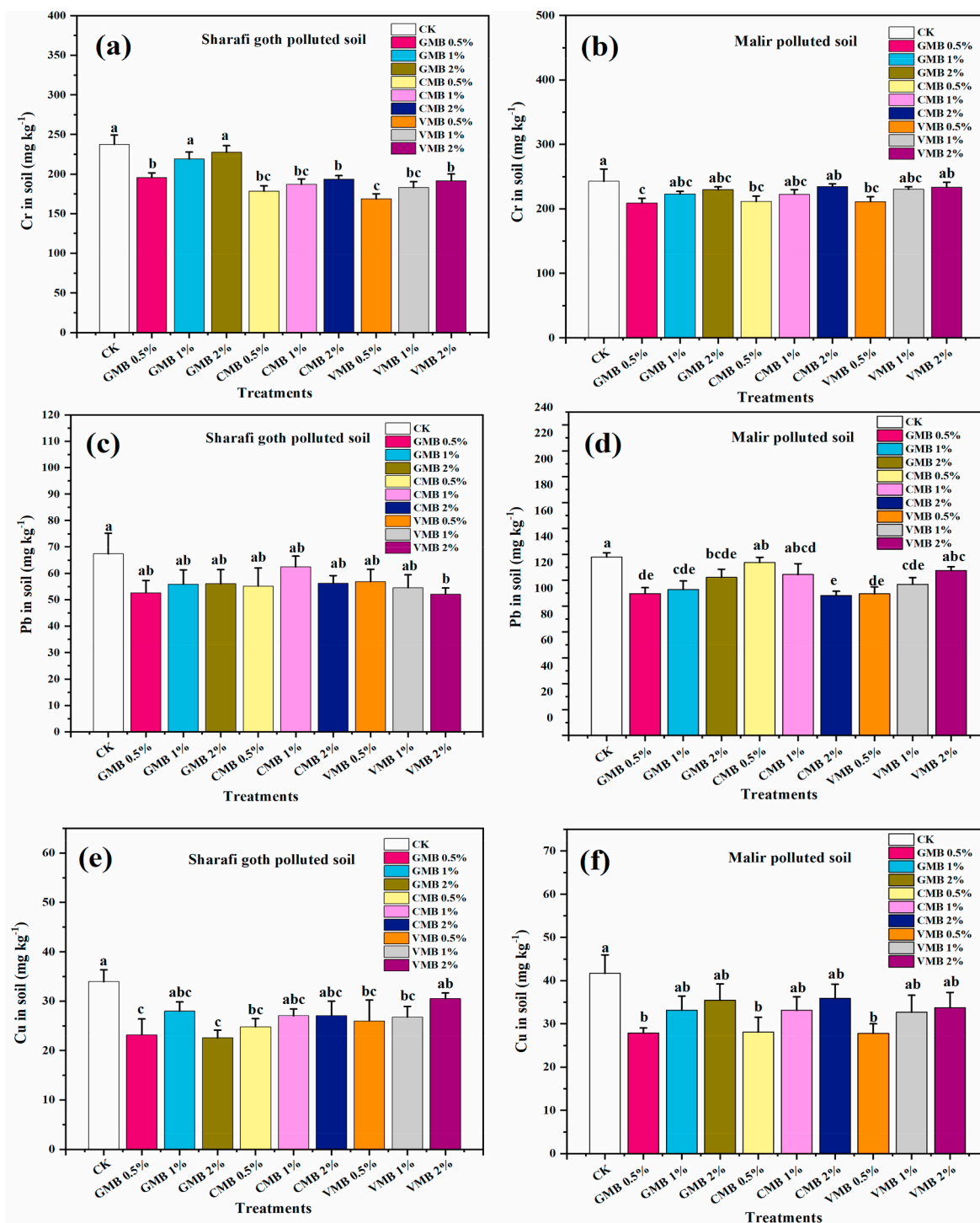


Fig. 4. Impact of modified biochars on: Cr in Sharafi goth polluted soil (a) and Malir polluted soil (b); Pb in Sharafi goth polluted soil (c) and Malir polluted soil (d); Cu in Sharafi goth polluted soil (e) and Malir polluted soil (f).

found that wheat plant growth was significantly increased following the application of wood BC in mining-polluted soil. However, Prapagdee et al. (2014) revealed that BC in a dose of 15% as a soil amendment caused opposite effects on plant growth.

3.6. Impact of rock phosphate-modified biochars on Cr, Pb, and Cu in polluted soils

The Cr, Pb, Cu, Zn, Ni, and Cd contents in Sharafi goth and Malir polluted soils were tested 50 days after mustard harvest. Cr in Sharafi Goth and Malir polluted soils was evidently immobilized, i.e. by 28.92% following the application of VMB 0.5% and by 13.92% after the application of GMB 0.5% as compared with control (Fig. 4a and b). The

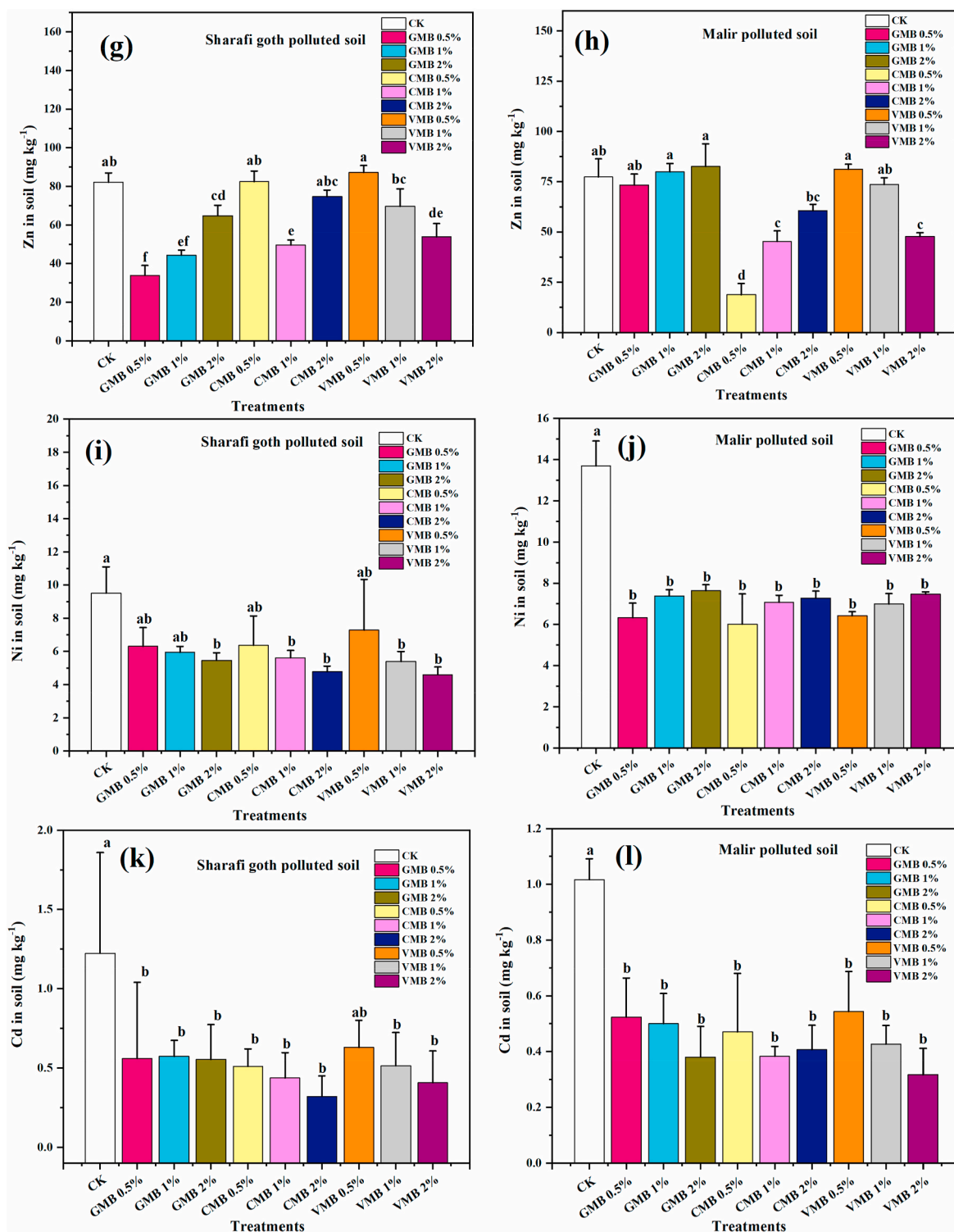


Fig. 5. Impact of modified biochars on: Zn in Sharafi goth polluted soil (g) and Malir polluted soil (h); Ni in Sharafi goth polluted soil (i) and Malir polluted soil (j); Cd in Sharafi goth polluted soil (k) and Malir polluted soil (l).

maximum immobilization of Pb in Sharafi goth polluted soil reached 22.65% with VMB 2%; similarly, the immobilization of Pb in Malir polluted soil was 21.63% with CMB 2% as compared with control (Fig. 4c and d). The Cu content in Sharafi goth polluted soil was significantly reduced to 33.55% with the application of GMB 2%. What is more, the Cu content in Malir polluted soil was substantially reduced to 33.3% after applying a low (0.5%) VMB dose compared to control

(Fig. 4e and f). Ali et al. (2019) discovered that Pb and Cu in mining soil were clearly immobilized after adding wood BC, which was due to the increase in soil pH. Arabi et al. (2021) observed the immobilization of Cr in acidic soil following the application of wood BC. Lu et al. (2017) reported that BCs from bamboo and rice straw had the potential to bind metals with organic matter in treated samples. Luo et al. (2022) reported that P-composite BC exhibited higher Pb and Cd fixation potential

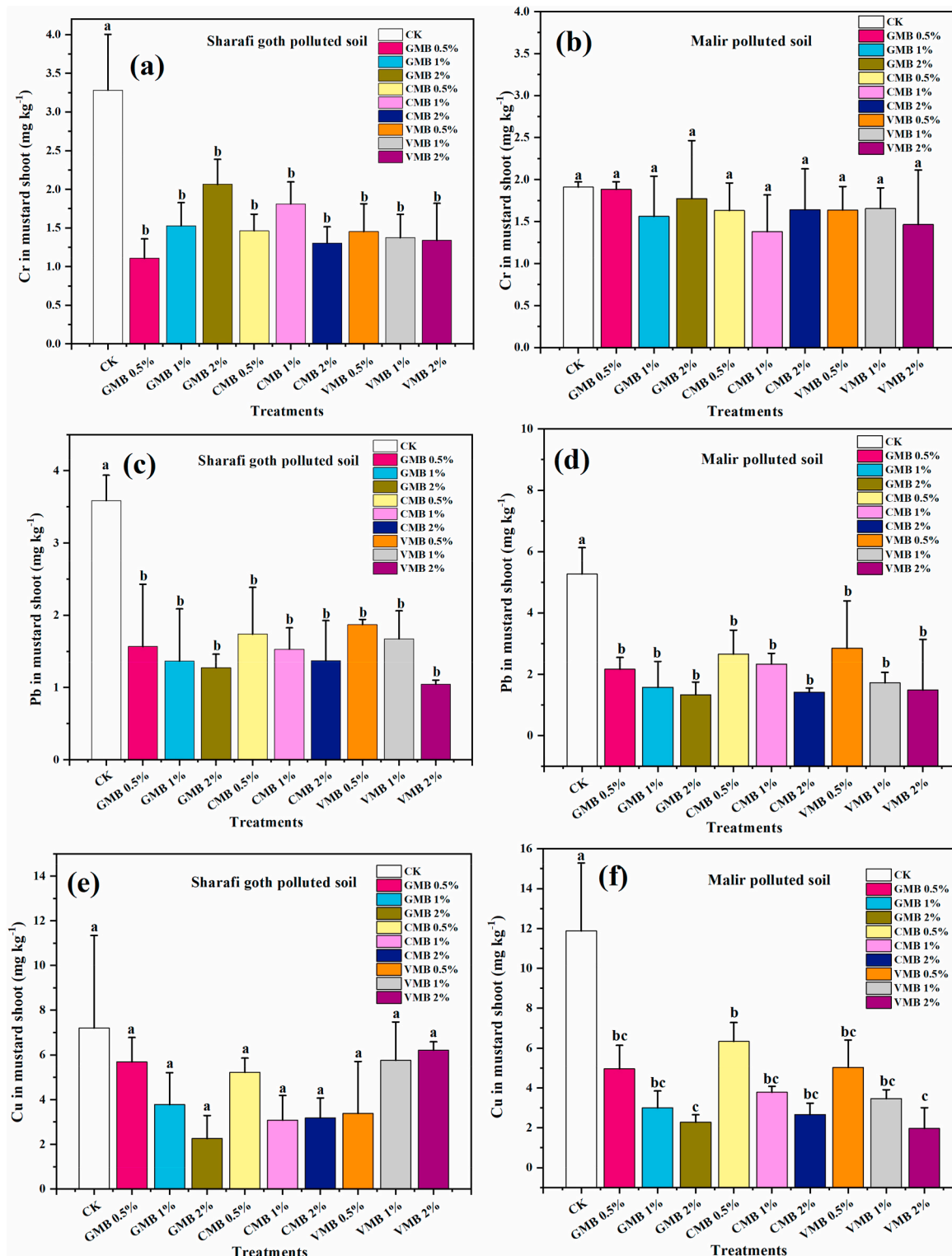


Fig. 6. Impact of modified biochars on: Cr in mustard shoots of Sharafi goth polluted soil (a) and Malir polluted soil (b); Pb in mustard shoots of Sharafi goth polluted soil (c) and Malir polluted soil (d); Cu in mustard shoots of Sharafi goth polluted soil (e) and Malir polluted soil (f).

(31.3–92.3%) than pristine BC (9.5–47.2%), which was mainly caused by the formation of stable precipitates such as $Pb_5(PO_4)_3Cl$ and $Cd_3(PO_4)_2$, particularly for the $Ca_5(PO_4)_3(OH)$ variant.

3.7. Impact of rock phosphate-modified biochars on Zn, Ni, and Cd in polluted soils

Compared to the control treatment, Zn in Sharafi goth polluted soil was significantly mobilized to 0.38% after the addition of CMB 0.5% and to 5.9% after applying VMB 0.5%, whereas the addition of GMB 0.5% caused immobilization of Zn by 58.73%. Also, Zn in Malir polluted soil was evidently mobilized to 3.15% with GMB 1%, 6.34% with GMB 2%, and 4.78% with VMB 0.5%; however, the maximum stabilization of Zn (75.73%) was observed after adding CMB 0.5%, as compared with control (Fig. 5g and h). The Ni content was significantly reduced, i.e. by 0.05%, after the application of VMB 2% to Sharafi goth polluted soil, and by 56.09% after the application of CMB 0.5% to Malir polluted soil, as compared with control (Fig. 5i and j). The maximum immobilization of Cd in Sharafi goth and Malir polluted soils reached 73.8% after the addition of CMB 2% and 68.7% after the addition of VMB 2% compared to control (Fig. 5k and l). It was observed that rock phosphate-modified biochars were more effective in immobilizing Cr, Pb, Cu, and Ni than Zn in Sharafi goth and Malir polluted soils, which may be due to the increase of OM in polluted soils. Liang et al. (2014) showed that the major mechanism underlying phosphate-induced PTE retention in soil was mainly due to metal-phosphate precipitation, while both sorption and precipitation were responsible for metal stabilization in BC addition. Puga et al. (2015) demonstrated that BC can retain huge proportions of PTEs in soil due to, among other things, its characteristics and large surface area. Ali et al. (2019) found that Zn and Cd in mining-polluted soil were markedly stabilized after the addition of wood BC, which may be caused by the increase in soil pH. However, Lu et al. (2017) reported that BC produced from rice straw and introduced at 5% evidently increased the organically bound Zn fraction in polluted soil. Arabi et al. (2021) stated that wood BC had the potential to immobilize Ni in acidic soils. Ge et al. (2022) found that the dominant fixation mechanism in their synergistic application was that H_3PO_4 -modified hydrochar (BPH) could fix Pb, Cd, and Cu by precipitation, complexation, and π - π electron-donor-acceptor interaction.

3.8. Impact of rock phosphate-modified-biochars on Cr, Pb, and Cu uptake by mustard shoots

Cr uptake by shoots was highly significantly reduced (by 66.15%) when GMB 0.5% was applied in Sharafi goth polluted soil, while a non-significant (27.74%) reduction in Cr uptake by mustard was observed following the addition of CMB 1% to Malir polluted soil (Fig. 6a and b). The maximum Pb uptake by mustard shoots was significantly reduced (by 60.7%) after applying VMB 2% in Sharafi goth polluted soil as compared to control. Additionally, the greatest reduction of Pb in mustard shoots was observed following the application of GMB 2% in Malir polluted soil, which was 74.76% compared to the control treatment (Fig. 6c and d). The application of GMB 2% as an amendment in Sharafi goth polluted soil resulted in the maximum decrease in Cu uptake by mustard shoots, accounting for 68.7% compared to control. The highest reduction (83.41%) in Cu uptake by mustard shoots was determined after applying VMB 2% in Malir polluted soil as compared to the treatment without additives (Fig. 6e and f). The study by Ali et al. (2019) revealed that Pb and Cu uptake by wheat shoots was clearly reduced after application of wood BC, which was due to increase in soil pH. Letsoalo (2020) found that BC from acacia and poultry litter considerably reduced Cr uptake by spinach.

3.9. Impact of rock phosphate-modified biochars on Zn, Ni, and Cd uptake by mustard shoots

The maximum increase in the Zn content in mustard shoots (14.84%) was observed with the application of GMB1%, 29.32% with CMB 0.5%, and 29.8% with VMB 0.5%. On the other hand, the greatest reduction in Zn in mustard shoots (28.21%) was noted after the addition of CMB 2% in Sharafi goth polluted soil, as compared with control. It was observed that GMB 1%, CMB 0.5%, and VMB 0.5% had the potential to increase Zn uptake by a hyperaccumulator plant, and the main reason for this mechanism was reduced soil pH. Soils in Pakistan are 70% deficient in Zn, especially land used for the cultivation of rice, vegetables, and cotton. These rock phosphate-modified biochars, introduced at low rates, can be successfully used to enhance Zn mobility in industrially polluted soil and increase Zn uptake by plants. However, it was found that the high-dose application of CMB 2% reduced the Zn content in polluted soil due to an almost 1-fold increase in soil pH (Fig. 7g). The maximum detected increase in Zn uptake by mustard shoots was: 1.8% with GMB 1%, 12.5% with GMB 2%, and 3.11% with CMB 0.5%; however, the application of VMB 2% reduced the Zn concentration in mustard shoots by 35% as compared with control. The above results indicated that GMB 1%, GMB 2%, and CMB substantially enhanced Zn uptake by plants as a result of lowering soil pH. Furthermore, VMB 2% in high dose reduced Zn uptake by the plant in Malir wastewater-treated soil, as soil pH increased up to 1.11 folds (Fig. 7h). Namgay et al. (2010) reported that the application of activated wood BC increased the concentration of Zn in maize plants.

Ni uptake by mustard shoots was significantly reduced to 86.36% following the application of GMB 2% in Sharafi goth polluted soil, and 76.44% reduction of Ni in mustard shoots was observed with the application of CMB 2% in Malir polluted soil, as compared to control (Fig. 7i and j). Compared to control, the maximum reduction of Cd concentration in mustard shoots was 89.41% after adding GMB 2% in Sharafi goth polluted soil; similarly, the maximum reduction of Cd in mustard shoots was 99.13% following the application of VMB 2% in Malir polluted soil (Fig. 7k and l). In this context, we hypothesized that MBs can stabilize PTEs in polluted soils, improve plant growth, and reduce PTE uptake by mustard. After introducing MBs into polluted soil, we compared the above results with our hypothesis and confirmed that PTEs were evidently immobilized Cr, Pb, Cu, Ni and Cd except Zn in soil-plant system, due to changes in soil chemical properties, including pH. Ali et al. (2019) applied wood BC in mining-polluted soil and observed that the accumulation of Zn and Cd in wheat shoots was significantly reduced with increasing pH. Letsoalo (2020) showed that BC from acacia and poultry litter significantly decreased Ni accumulation in spinach.

3.10. Redundancy analysis of the studied parameters

According to the obtained data, the relationships between EC, pH, OM, and SOC are indicated by red arrows and between fresh and dry biomass of mustard, Zn, Cd, Ni, Cr, Pb, and Cu in soil and maize shoots are indicated by blue arrows (Fig. 8a). RDA revealed that the studied parameters can explain (74.25%) of the total variance in Sharafi goth polluted soil. Soil pH was significantly negatively correlated with Zn in soil and mustard shoots. Similarly, soil EC, OM, SOC, and pH were significantly negatively correlated with Cu, Ni, Cd, Cr, and Pb in mustard plants and Sharafi goth polluted soil. In addition, EC showed a negative correlation with mustard fresh and dry biomasses. The data obtained for Malir polluted soil revealed that the studied parameters can explain (52.65%) of the total variance. The data showed that OM was significantly positively correlated with EC, pH, and Zn uptake in mustard shoots; however, a negative correlation of OM was observed with lead, cadmium, zinc, nickel, copper, and chromium in Malir polluted soil. Moreover, soil EC and pH were significantly negatively correlated with lead, cadmium, zinc, nickel, copper, and chromium in

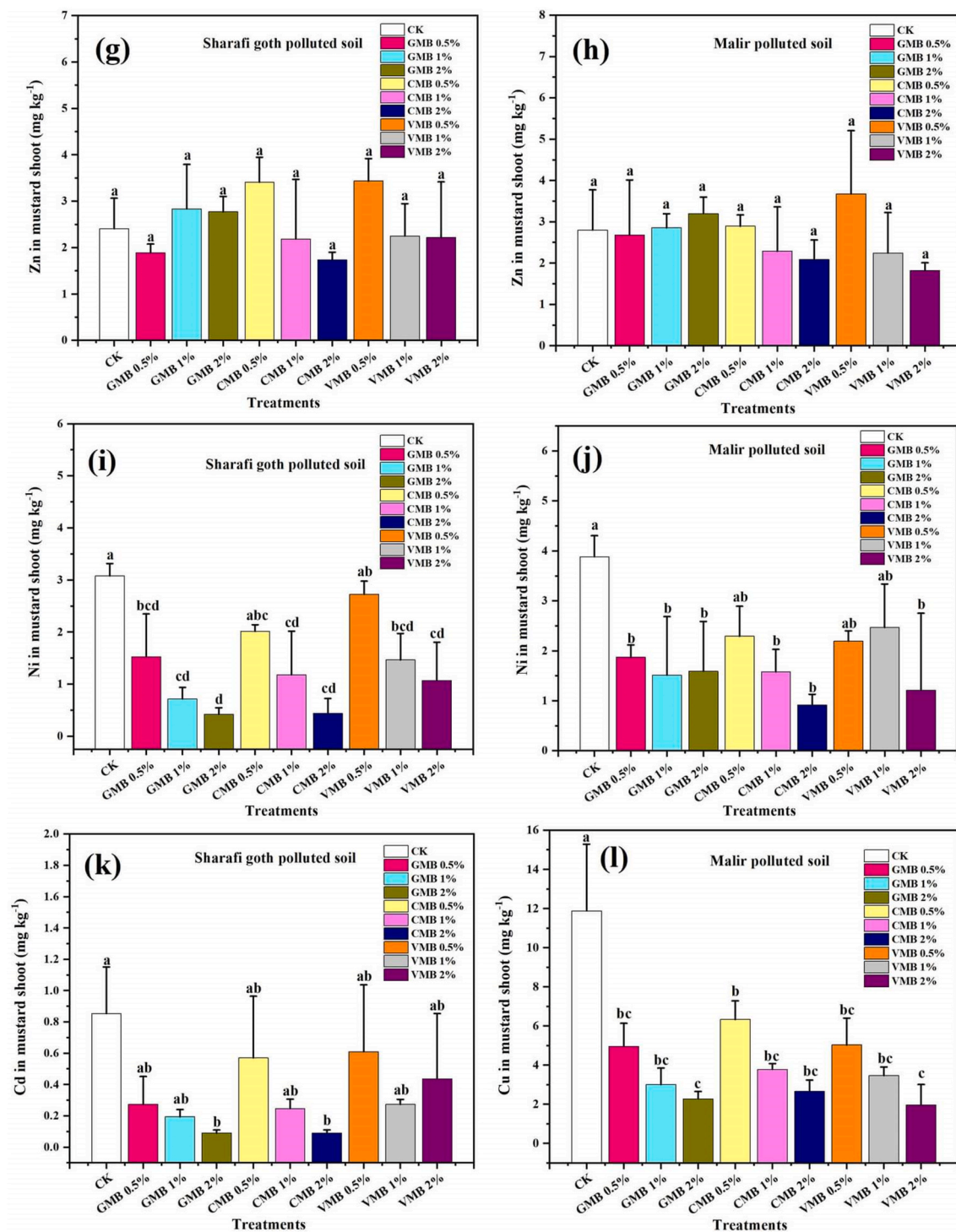


Fig. 7. Impact of modified biochars on: Zn in mustard shoots of Sharafi goth polluted soil (g) and Malir polluted soil (h); Ni in mustard shoots of Sharafi goth polluted soil (i) and Malir polluted soil (j).

Malir polluted soil and mustard shoots (Fig. 8b). A.H. Lahori et al. (2020) performed a redundancy analysis between PTEs and soil chemical properties using minerals and biochar. The obtained results indicated that the mobility and immobility of PTEs depend on soil chemical properties, soil type, and nature of additives.

4. Conclusions

The main objective of the present study was to assess the

effectiveness of three different rock phosphate-modified BCs introduced in 0.5, 1, and 2% doses on PTE fixation in Sharafi goth and Malir polluted soils. The results indicated that modified BCs GMB 0.5%, GMB 1%, and GMB 2% increased the fresh biomass of mustard, but VMB 0.5% reduced the fresh biomass of mustard grown in Sharfi goth polluted soil. In addition, GMB 0.5%, GMB 1%, GMB 2%, VMB 0.5%, and VMB 2% potentially enhanced the fresh biomass of mustard, while VMB 1% reduced the fresh biomass of mustard in Malir polluted soil compared to control. Application of MBs potentially immobilized Cr, Pb, Cu, Ni, and

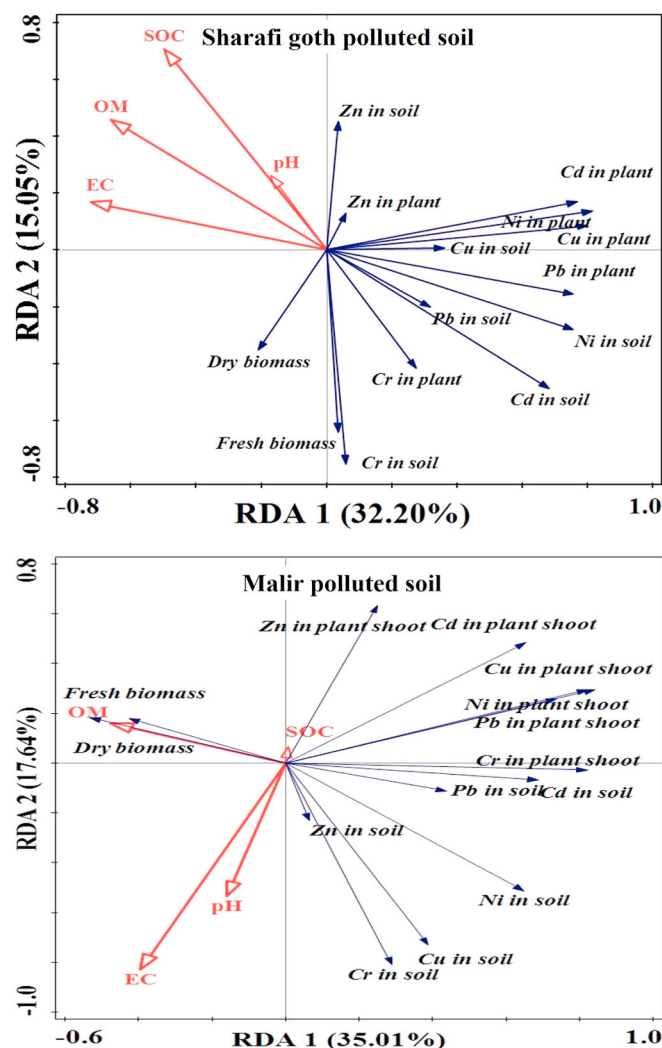


Fig. 8. Redundancy analysis of the studied parameters in polluted soils.

Cd in Sharafi goth and Malir polluted soils. However, CMB 0.5% and VMB 0.5% mobilized Zn in Sharafi goth polluted soil and GMB 1%, GMB 2%, and VMB 0.5% as amendments mobilized Zn in Malir wastewater-polluted soil. The uptake of Cr, Pb, Cu, Ni, and Cd by mustard shoots was significantly reduced; however, GMB 1%, CMB 0.5%, and VMB 0.5% treatments increased Zn uptake in Sharafi goth polluted soil and GMB 1%, GMB 2%, CMB 0.5%, and VMB 0.5% treatments caused the same in Malir polluted soil. This may be due to changes in EC, pH, OM, and SOC following the application of BCs. Soils in Pakistan are 70% deficient in Zn, especially land used for the cultivation of rice, vegetables, and cotton. It was further concluded that 0.5–1% BCs doses can be successfully used in Zn deficient areas, but 2% dose can stabilize Zn toxicity in polluted soils. Overall, the present study recommends long-term field experiments to verify the residual impact of MBs at different doses and particle sizes on the dominant mechanism of PTE fixation in polluted soils under various field crops. Future studies need to be carried out with BCs produced at different pyrolysis temperatures, in various doses, combined with other materials such as nanomaterials, minerals, etc. to evaluate their effect on soil enzymatic activity, macro-/micronutrient contents, molecular level, and stability of organic/inorganic contaminants from polluted areas.

Author contributions

Tanveer Hussain and Altaf Hussain Lahori: Conceptualization,

Methodology and Writing – original draft; Monika Mierzwa-Hersztek: Investigation, Writing – review & editing; Samreen Riaz Ahmed and Viola Vambol: Investigation, Writing – review & editing; Muhammad Faizan Shahid: Writing – review & editing and Software; Altaf Hussain Lahori: Software and Validation; Asif Ali Khan: Writing – review & editing; Lubna Rafique: Formal analysis; Sajid wasia and Zhang Zengqiang: Formal analysis.

Conceptualization

Tanveer Hussain and Altaf Hussain Lahori.; **Data curation**; Tanveer Hussain.; **Formal analysis**; Muhammad Faizan Shahid.

Funding acquisition

Altaf Hussain Lahori.; **Investigation**; Samreen Riaz Ahmed and Viola Vambol.; **Methodology**; Tanveer Hussain, Altaf Hussain Lahori and Monika Mierzwa-Hersztek.; **Project administration**; Altaf Hussain Lahori.; **Resources**; Altaf Hussain Lahori.; **Software**; Altaf Hussain Lahori and Tanveer Hussain.; **Supervision**; Altaf Hussain Lahori.; **Validation**; Asif Ali Khan.; **Visualization**; Sajid Wasia and Zhang Zengqiang.; **Roles/Writing - original draft**; Altaf Hussain Lahori and Tanveer Hussain.; **Writing - review & editing**. Monika Mierzwa-Hersztek and Lubna Rafique.

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Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

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